The number of outermost (valence) electrons an element has determines which group of the periodic table the element is placed in. For example, if an element has three electrons in its outermost shell then it will be placed in Group III. Without looking at a periodic table, infer which group these elements belong to.

Write your answers in the table below.

<table>
<thead>
<tr>
<th>Group I</th>
<th>Group II</th>
<th>Group III</th>
<th>Group IV</th>
<th>Group V</th>
<th>Group VI</th>
<th>Group VII</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
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